# One step synthesis and densification of nanocrystalline $\mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathbf{N}_{4}$ composite from mechanically activated powders by high-frequency induction-heated combustion 

Hyun-Kuk Park • Jeong-Hwan Park • Jin-Kook Yoon •<br>Ki-Tae Lee • In-Jin Shon

Received: 28 May 2007 / Accepted: 18 June 2008 /Published online: 9 July 2008
(C) Springer Science + Business Media, LLC 2008


#### Abstract

Dense nanocrystalline $4 \mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite was synthesized by high-frequency induction-heated combustion synthesis (HFIHCS) method within 1 min in one step from mechanically activated powders of TaN and Si . Simultaneous combustion synthesis and densification were accomplished under the combined effects of an induced current and mechanical pressure. Highly dense $4 \mathrm{TaSi}_{2}{ }^{-}$ $\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite with relative density of up to $99 \%$ was produced under simultaneous application of a 60 MPa pressure and the induced current. The average grain size and mechanical properties (hardness and fracture toughness) of the composite were investigated.


Keywords High-frequency induction heated combustion • Composite materials • Sintering • Nanophase • Mechanical properties $\cdot \mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$

## 1 Introduction

High melting point intermetallic compounds with low density and improved oxidation resistance have received wide attention as potential aerospace materials over the past

[^0]20 years. In this regard, transition-metal silicides are very attractive for application temperature up to $1300{ }^{\circ} \mathrm{C}$ and higher because this class of materials has an attractive combination of properties, including high melting temperature, high modulus, high oxidation resistance in air, and a relatively low density $[1,2]$. In addition, the thermal and electrical conductivities are relatively high and therefore they are also attractive for electronic interconnections and diffusion barriers. Most of the investigations on silicides have focused on a few compounds, with $\mathrm{MoSi}_{2}$ and $\mathrm{TiSi}_{2}$ being the most studied. Other silicides, such as $\mathrm{TaSi}_{2}$, $\mathrm{WSi}_{2}, \mathrm{NbSi}_{2}, \mathrm{ZrSi}_{2}$ and $\mathrm{VSi}_{2}$ have received relatively little attention. The melting point, crystal structure, density, formation enthalpy at 298 K , and adiabatic temperature of $\mathrm{TaSi}_{2}$ are $2025^{\circ} \mathrm{C}$, hexagonal, $9.205 \mathrm{~g} / \mathrm{cm}^{3}, 32.4 \mathrm{~kJ} \mathrm{~mol}^{-1}$, and $1,795 \mathrm{~K}[3-5]$. As in the case of many intermetallic compounds, the current concern about these materials focuses on their low fracture toughness below the ductilebrittle transition temperature [6-8]. To improve on their mechanical properties, the approach commonly utilized has been the addition of a second phase to form composites [914]. An example is the addition of $\mathrm{Si}_{3} \mathrm{~N}_{4}$ to $\mathrm{TaSi}_{2}$ to improve the latter's properties. Silicon nitride has a high thermal shock resistance, due to its low thermal expansion coefficient, and a good resistance to oxidation when compared to other structural materials [15, 16]. The isothermal oxidation resistance of $\mathrm{NbSi}_{2}-40$ vol $\% \mathrm{Si}_{3} \mathrm{~N}_{4}$ composite prepared by spark plasma sintering (SPS) process in dry air at $1300{ }^{\circ} \mathrm{C}$ was superior to that of monolithic $\mathrm{NbSi}_{2}$ compact since the composite contained a larger amount of Si , which made it easier to form dense $\mathrm{SiO}_{2}$ scale [17]. Therefore, $\mathrm{Si}_{3} \mathrm{~N}_{4}$ may be the most promising additive as a reinforcing material for $\mathrm{TaSi}_{2}$-based composites.

Many similar high-temperature dense composites are usually prepared in a multi-step process [18, 19]. However, the method of field-activated and pressure-assisted combustion synthesis has been successfully employed to synthesize and densify materials from the elements in one step in a relatively short period of time. This method has been used to synthesize a variety of ceramics and composites, including $\mathrm{MoSi}_{2}-\mathrm{ZrO}_{2}, \mathrm{Ti}_{5} \mathrm{Si}_{3}$ and its composites, $\mathrm{WSi}_{2}$ and its composites, and WC-Co hard materials [20-25]. These materials, which are generally characterized by low adiabatic combustion temperature, cannot be synthesized directly by the self-propagating high-temperature synthesis (SHS) method. More recently, a new approach has been developed in which synthesis and densification can be effected simultaneously. This new process, referred to as the high-frequency induction-heated combustion synthesis (HFIHCS), has been successfully used to synthesize and densify materials in one step in a relatively short period of time ( 2 min ) [26-28].

The objective of this study is to investigate the preparation of dense nanocrystalline $4 \mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite by the HFIHCS method starting from a mixture of mechanically activated TaN and Si powders. The interaction between these phases, i.e.,
$4 \mathrm{TaN}+11 \mathrm{Si} \rightarrow 4 \mathrm{TaSi}_{2}+\mathrm{Si}_{3} \mathrm{~N}_{4}$
is thermodynamically feasible, as can be seen in Fig. 1.


Fig. 1 Temperature dependence of the Gibbs free energy for the interaction between TaN and silicon


Fig. 2 Scanning election microscope images of raw materials: (a) tantalum nitride, (b) silicon

## 2 Experimental procedure

Powders of $99.5 \%$ tantalum nitride ( -325 mesh, Alfa Products, Ward Hill, MA) and 99.5\% pure silicon (-325 mesh, Alfa Products, Ward Hill, MA) were used as starting materials. Figure 2 shows the SEM images of the raw materials used. Powder mixtures of TaN and Si in the molar proportion of $4: 11$ were first milled in a high-energy ball mill (Pulverisette-5, planetary mill) at 250 rpm for 10 h . Tungsten carbide balls ( 5 mm in diameter) were used in a sealed cylindrical stainless steel vial under argon atmosphere. The weight ratio of ball-to-powder was 30:1. Milling resulted in a significant reduction of grain size. The grain size and the internal strain were determined by XRD analysis.

After milling, the mixed powders were placed in a graphite die (outside diameter, 45 mm ; inside diameter, 20 mm ; height, 40 mm ) and then introduced into the highfrequency induction-heated combustion system, shown schematically in Fig. 3. Following the introduction of the die into the apparatus, the system was evacuated and a


Fig. 3 Schematic diagram of the high-frequency induction heated combustion apparatus
uniaxial pressure of 60 MPa was applied. An induced current (frequency of about 50 kHz ) was then activated and maintained until densification was attained as indicated by a linear gauge measuring the shrinkage of the sample. Temperatures were measured by a pyrometer focused on the surface of the graphite die. At the end of the process, the sample was cooled to room temperature. The process was carried out under a vacuum of 40 mTorr .

The relative densities of the synthesized sample were measured by the Archimedes method. Microstructural characterization was made on product samples which had been polished and etched using a solution of HF ( $30 \mathrm{vol} . \%$ ), $\mathrm{HNO}_{3}$ ( $30 \mathrm{vol} . \%$ ) and $\mathrm{H}_{2} \mathrm{O}(40 \mathrm{vol} . \%)$ for 8 s at room temperature. Compositional and microstructural analyses of the products were made through X-ray diffraction (XRD) and scanning electron microscopy (SEM) with energy dispersive X-ray analysis (EDAX). Vickers hardness was measured by performing indentations at a load of 10 kg and a dwell time of 15 s .

## 3 Results and discussion

Figure 4 shows XRD patterns of the raw powders and the milled $4 \mathrm{TaN}+11 \mathrm{Si}$ powder mixture. The FWHM of the


Fig. 4 XRD patterns of raw materials: (a) TaN , (b) Si and (c) milled $4 \mathrm{TaN}+11 \mathrm{Si}$


Fig. 5 Variations of temperature and shrinkage displacement with heating time during high-frequency induction heated combustion synthesis and densification of $4 \mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite (under $60 \mathrm{MPa}, 90 \%$ output of total power capacity)
milled powder is greater than that of the raw powders due to internal strain and grain size reduction. The grain size and the internal strain can be calculated by Stoke and Wilson's formula [29],
$b=b_{\mathrm{d}}+b_{\mathrm{e}}=k \lambda /(d \cos \theta)+4 \varepsilon \tan \theta$
where $b$ is the full width at half-maximum (FWHM) of the diffraction peak after instrument correction; $b_{\mathrm{d}}$ and $b_{\mathrm{e}}$ are FWHM caused by small grain size and internal strain, respectively; $k$ is constant (with a value of 0.9 ); $\lambda$ is wavelength of the X-ray radiation; $d$ and $\varepsilon$ are grain size and internal strain, respectively; and $\theta$ is the Bragg angle. The parameters $b$ and $b_{s}$ follow Cauchy's form with the relationship: $B_{0}=b+b_{s}$, where $B_{0}$ and $b_{s}$ are FWHM of the broadened Bragg peaks and the standard sample's Bragg peaks, respectively.

The calculated average grain sizes and internal strains of the milled TaN and Si powders are 68 and $37 \mathrm{~nm}, 0.004$ and 0.002 , respectively.

The variations in shrinkage displacement and temperature with heating time during the processing of $4 \mathrm{TaN}+11 \mathrm{Si}$ system are shown in Fig. 5. As the induced current was applied, the shrinkage displacement increased gradually with temperature up to about $950{ }^{\circ} \mathrm{C}$, but then abruptly
increased at about $1000{ }^{\circ} \mathrm{C}$. When the reactant mixture of $4 \mathrm{TaN}+11 \mathrm{Si}$ was heated under 60 MPa pressure to $950{ }^{\circ} \mathrm{C}$, no reaction took place and no significant shrinkage displacement as judged by subsequent XRD and SEM analyses. Figure 6(a), (b), and (c) show the SEM (second-


Fig. 6 Scanning electron microscope images of $4 \mathrm{TaN}+11 \mathrm{Si}$ system: (a) after milling (b) heating at $950{ }^{\circ} \mathrm{C}$ before combustion synthesis and (c) heating at $1100^{\circ} \mathrm{C}$ after combustion synthesis


Fig. 7 XRD patterns of the 4TaN + 11Si system: (a) after milling (b) heating at $950{ }^{\circ} \mathrm{C}$ before combustion synthesis and (c) heating at $1100{ }^{\circ} \mathrm{C}$ after combustion synthesis
ary electron) images of (a) powder after milling, (b) sample heated to $950{ }^{\circ} \mathrm{C}$ and (c) sample heated to $1100{ }^{\circ} \mathrm{C}$, respectively. Figures 6(a) and (b) show the presence of the reactants as separate phases. X-ray diffraction results shown in Fig. 7(a) and (b) exhibit only peaks pertaining to the reactants of TaN and Si . However, when the temperature was raised to $1100{ }^{\circ} \mathrm{C}$, the starting powders react and then form highly dense products. SEM image of an etched surface of the samples heated to $1100^{\circ} \mathrm{C}$ under a pressure of 60 MPa is shown in Fig. 6(c). A complete reaction between TaN and Si took place under these conditions. X-ray diffraction analyses of this sample showed peaks of only $\mathrm{TaSi}_{2}$ and $\mathrm{Si}_{3} \mathrm{~N}_{4}$, as indicated in Fig. 7(c). The abrupt increase in the shrinkage displacement at the ignition temperature is due to the increase in density as a result of molar volume change associated with the formation of $4 \mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ from the reactants ( TaN and Si ) and the consolidation of the product. It should be recalled that the measured temperatures are those of the surface of the die and are, therefore, likely to be different than the values in the middle of the sample. Thus, the onset of the reaction to form the composite (and the concomitant rapid shrinkage) may be at a higher temperature than the observed value of $950{ }^{\circ} \mathrm{C}$

The average grain sizes of $\mathrm{TaSi}_{2}$ and $\mathrm{Si}_{3} \mathrm{~N}_{4}$ calculated by the Stoke-Wilson's formula [29] were about 80 and 60 nm . The $\mathrm{Si}_{3} \mathrm{~N}_{4}$ particles were well distributed in matrix, as can bee seen from the SEM image, Fig. 6(c).

Vickers hardness measurements were made on polished sections of the $\mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite using a 10 kg load and 15 s dwell time. The calculated hardness value, based on an average of five measurements, of the $\mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite is $1,403 \pm 26 \mathrm{~kg} / \mathrm{mm}^{2}$. Indentations with large
enough loads produced median cracks around the indent. From the length of these cracks, fracture toughness values can be determined using two expressions. The first expression, proposed by Anstis et al. [30] is
$K_{\text {IC }}=0.016(E / H)^{1 / 2} \cdot P / C^{3 / 2}$
where $E$ is Young's modulus, H the indentation hardness, $P$ the indentation load, and $C$ the trace length of the crack measured from the center of the indentation, the modulus was estimated by the mixtures rule of the 0.129 volume fraction of $\mathrm{Si}_{3} \mathrm{~N}_{4}$ and the 0.871 volume fraction of $\mathrm{TaSi}_{2}$ using $E\left(\mathrm{Si}_{3} \mathrm{~N}_{4}\right)=308 \mathrm{GPa}$ [31] and $E\left(\mathrm{TaSi}_{2}\right)=357 \mathrm{GPa}$ [5]. The second expression, proposed by Niihara et al. [32, 33], is
$K_{\text {IC }}=0.023(c / a)^{-3 / 2} \cdot H_{v} \cdot a^{1 / 2}$
where $c$ is the trace length of the crack measured from the center of the indentation, $a$ the half of average length of two indent diagonals, and $H_{\mathrm{v}}$ the hardness.

As in the case of hardness values, the toughness values were derived from the average of five measurements. The


Fig. 8 (a) Vickers hardness indentation and (b) median crack propagating of $4 \mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite
toughness values obtained by the two methods of calculation are $2.9 \pm 0.3$ and $3.4 \pm 0.3 \mathrm{MPam}^{1 / 2}$, respectively.

A typical indentation pattern for the $4 \mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite is shown in Fig. 8(a). Typically, one to three additional cracks were observed to propagate from the indentation corner. Higher magnification view of the indentation median crack in the composite is shown in Fig. 8(b). This shows the crack propagates along phase boundary of $\mathrm{TaSi}_{2}$ and $\mathrm{Si}_{3} \mathrm{~N}_{4}$.

Although the absence of reported values for hardness and toughness on $\mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ precludes making direct comparison to the results obtained in this work to show the influence of grain size, we can evaluate the mechanical properties from the analogous metal silicide- $\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite system. For example, Murakami et al. have reported that the addition of $\mathrm{Si}_{3} \mathrm{~N}_{4}$ to $\mathrm{NbSi}_{2}$ or $\mathrm{Nb}_{5} \mathrm{Si}_{3}$ matrix increases the microvickers hardness [17]. We may analogy that the $\mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite may show lower hardness compared to $\mathrm{Si}_{3} \mathrm{~N}_{4}$. The further study, however, may be necessary to verify the effect of grain size and the volume fraction of each component on mechanical properties in the $\mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite system.

## 4 Summary

Using the high-frequency induction-heated combustion method, the simultaneous synthesis and densification of nanocrystalline $4 \mathrm{TaSi}_{2}-\mathrm{Si}_{3} \mathrm{~N}_{4}$ composite was accomplished using powders of TaN and Si. Complete synthesis and densification can be achieved in one step within 1 min . The relative density of the composite was $99 \%$ under an applied pressure of 60 MPa and the induced current. The average grain sizes of $\mathrm{TaSi}_{2}$ and $\mathrm{Si}_{3} \mathrm{~N}_{4}$ phases in the composite were about 80 and 60 nm , respectively. The average hardness and fracture toughness values obtained were $1,403 \mathrm{~kg} / \mathrm{mm}^{2}$ and $3.4 \mathrm{MPa} \cdot \mathrm{m}^{1 / 2}$, respectively.

## References

1. A.K. Vasudevan, J.J. Petrovic, J. Mater. Sci. Eng A155, 259 (1992) doi:10.1016/0921-5093(92)90308-N
2. G.J. Fan, M.X. Quan, Z.Q. Hu, J. Eckert, L. Schulz, Scripta Mater 41, 1147 (1999) doi:10.1016/S1359-6462(99)00285-7
3. H. Okamoto, Phase diagrams for binary alloys (ASM international, 2000), pp 735
4. O. Knacke, O. Kubaschewski, K. Hesselmann, Springer-Verlag, 1973 (1991)
5. F. Chu, M. Lei, S.A. Maloy, J.J. Petrovic, T.E. Mitchell, Acta mater 44, 3035 (1966) doi:10.1016/1359-6454(95)00442-4
6. G. Sauthoff, Intermetallics (VCH Publishers, New York, 1995)
7. Y. Ohya, M.J. Hoffmann, G. Petzow, J. Mater. Sci. Lett 12, 149 (1993) doi:10.1007/BF00819942
8. J. Qian, L.L. Daemen, Y. Zhao, Diam. \& Related Mater 14, 1669 (2005) doi:10.1016/j.diamond.2005.06.007
9. B.W. Lin, T. Iseki, Br. Ceram. Trans. J 91, 1 (1992)
10. Y. Ohya, M.J. Hoffmann, G. Petzow, J. Am. Ceram. Soc 75, 2479 (1992) doi:10.1111/j.1151-2916.1992.tb05600.x
11. S.K. Bhaumik, C. Divakar, A.K. Singh, G.S. Upadhyaya, J. Mater. Sci. Eng A279, 275 (2000) doi:10.1016/S0921-5093(99)00217-8
12. D.K. Jang, R. Abbaschian, Kor. J. Mater. Res 9, 92 (1999)
13. H. Zhang, P. Chen, M. Wang, X. Liu, Rare Metals 21, 304 (2002)
14. D.Y. Oh, H.C. Kim, J.K. Yoon, I.J. Shon, Alloys \& Compounds 395, 174 (2005) doi:10.1016/j.jallcom.2004.10.072
15. W. Dressler, R. Riedel, Int. J. Refract. Met. Hard Mater 15, 13 (1997) doi:10.1016/S0263-4368(96)00046-7
16. S.P. Taguchi, S. Ribeiro, J. Mater. Proce. Tech 147, 336 (2004) doi:10.1016/j.jmatprotec.2004.01.003
17. T. Murakami, S. Sasaki, K. Ichikawa, A. Kitahara, Intermetallics 9, 621 (2001) doi:10.1016/S0966-9795(01)00042-5
18. K. Bhattacharya, J.J. Petrovic, J. Am. Ceram. Soc 74, 2700 (1991) doi:10.1111/j.1151-2916.1991.tb06828.x
19. Y. Luo, S. Li, W. Pan, L. Li, Mater. Lett 58, 150 (2003) doi:10.1016/S0167-577X(03)00434-8
20. Z.A. Munir, I.J. Shon, K. Yamazaki, U. S. Patent No. 5,794,113 (1998)
21. I.J. Shon, Z.A. Munir, K. Yamazaki, K. Shoda, J. Am. Ceram. Soc 79, 1875 (1996) doi:10.1111/j.1151-2916.1996.tb08008.x
22. I.J. Shon, H.C. Kim, D.H. Rho, Z.A. Munir, Mater. Sci. Eng A269, 129 (1999) doi:10.1016/S0921-5093(99)00131-8
23. I.J. Shon, D.H. Rho, H.C. Kim, Z.A. Munir, J. Alloys \& Compounds 322, 120 (2001) doi:10.1016/S0925-8388(01)01167-7
24. I.J. Shon, D.H. Rho, H.C. Kim, Met. \& Mater 6, 533 (2000)
25. C.D. Park, H.C. Kim, I.J. Shon, Z.A. Munir, J. Am. Ceram. Soc 85, 2670 (2002) doi:10.1111/j.1151-2916.2002.tb00492.x
26. H.C. Kim, D.Y. Oh, I.J. Shon, Int. J. Refrac. Met. \& Hard Mater 22, 41 (2004) doi:10.1016/j.ijrmhm.2003.12.002
27. H.C. Kim, D.Y. Oh, J. Guojian, I.J. Shon, Mater. Sci. Eng A368, 10 (2004) doi:10.1016/j.msea.2003.08.105
28. D.Y. Oh, H.C. Kim, J.K. Yoon, I.J. Shon, J. Alloys \& Compounds 395, 270 (2005) doi:10.1016/j.jallcom.2004.05.069
29. F.L. Zhang, C.Y. Wang, M. Zhu, Scripta Mater 49, 1123 (2003) doi:10.1016/j.scriptamat.2003.08.009
30. G.R. Anstis, P. Chantikul, B.R. Lawn, D.B. Marshall, J. Am. Ceram. Soc 64, 533 (1981) doi:10.1111/j.1151-2916.1981.tb10320.x
31. M. Lugovy, V. Slyunyayev, V. Subbotin, N. Orlovskaya, G. Gogotsi, Compos. Scie. and Tech 64, 1947 (2004) doi:10.1016/j. compscitech.2004.02.007
32. N. Koichi, Ceramics 20, 1218 (1985)
33. D.Y. Oh, H.C. Kim, J.K. Yoon, I.J. Shon, J. Alloys \& Compound 395, 174 (2005) doi:10.1016/j.jallcom.2004.10.072

[^0]:    H.-K. Park • J.-H. Park • K.-T. Lee • I.-J. Shon ( $\triangle$ )

    Division of Advanced Materials Engineering, Research Center of Industrial Technology, Chonbuk National University, 664-14 Deokjin-dong 1-ga, Deokjin-gu, Jeonju, Jeonbuk 561-756, Republic of Korea
    e-mail: ijshon@chonbuk.ac.kr
    J.-K. Yoon

    Advanced Functional Materials Research Center, Korea Institute of Science and Technology, P.O Box 131, Cheongryang, Seoul 130-650, Republic of Korea.

